## SHORT COMMUNICATIONS

EESTI NSV TEADUSTE AKADEEMIA TOIMETISED. KEEMIA ИЗВЕСТИЯ АКАДЕМИИ НАУК ЭСТОНСКОЙ ССР. ХИМИЯ PROCEEDINGS OF THE ACADEMY OF SCIENCES OF THE ESTONIAN SSR. CHEMISTRY

1985, 34, 3

УДК 539.192

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## THE ANTISYMMETRIC STRETCHING FREQUENCY OF BO<sub>2</sub>

- P.-E. LOMP, N. STEPANOV, V. SIMKIN, A. DEMENTIEV. BO: MOLEKULI ANTISÜMMEETRI-LISE VÕNKUMISE SAGEDUS
- П.-Э. ЛОМП, Н. СТЕПАНОВ, В. СИМКИН, А. ДЕМЕНТЬЕВ. ЧАСТОТА АНТИСИММЕТРИЧ-НОГО КОЛЕБАНИЯ МОЛЕКУЛЫ ВО2

### (Presented by M. Veiderma)

Boron dioxide (BO<sub>2</sub>) has been a subject of several experimental investigations; only a few theoretical ones have been carried out  $[^{1-3}]$ . In  $[^2]$  the  $C^2 \sum_g^+$  state was calculated, the results of perturbational CI technique showing good agreement with experimental data. In  $[^3]$  the results of *ab initio* SCF and CI calculations on doublet and quartet states of BO<sub>2</sub> have been represented, which confirm the linearity of the molecule in the ground  $X^2 \sum_g^+$  and low-lying excited states  $A^2 \Pi_u$ ,  $B^2 \sum_u^+$ and  $C^2 \sum_g^+$ . The  $v_1$  and  $v_2$  fundamentals were also calculated.

Here we present the results obtained for the  $v_3$  fundamental frequency.

Because of selection rules, direct determination of the value of antisymmetric stretching frequency  $v_3$  is quite complicated. J. W. C. Johns

[4] found the value  $\frac{1}{2}(2v_3) = 1322$  cm<sup>-1</sup>, L. V. Serebrennikov,

V. A. Koryazhkin et al. [5-7] obtained the  $v_3$  band in the region 1840— 2080 cm<sup>-1</sup>, and with the help of tunable infrared diode laser spectroscopy K. Kawaguchi et al. [8] found  $v_3 = 1278$  cm<sup>-1</sup>. As compared with analogous systems the  $v_3$  value in the 1300 cm<sup>-1</sup> region should be considered as anomalously low, and in [8] this was assigned to a vibronic interaction between  $X^2 \Pi_g$  and  $A^2 \Pi_u$  states.

To obtain the  $v_3$  value, *ab initio* restricted SCF, CASSCF and CI calculations were carried out in  $C_s$  and  $C_{\infty v}$  symmetries using the SPUSH program system [9]. The basis sets were the standard STO-3G set (basis I) [10] and (9s5p/4s2p) (basis II) [11]. From the reference set 200 singly and doubly excited configurations were selected [12], and the CI problem was solved with MOs from BO<sub>2</sub> molecule and BO<sub>2</sub><sup>-</sup> and

Total energies and antisymmetric

$\begin{array}{c} \Delta = R_1 - R_2 \\ (R_1, R_2)  \text{a. u} \end{array}$	1	2	3	4	5
	n wantil a wa	inacian a stran	na in part	Ven vrea	Basis I
$\begin{array}{c} 0.00\\ (2.47, 2.47)\\ 0.04\\ (2.45, 2.49)\\ 0.12\\ (2.41, 2.53)\\ 0.20\\ (2.37, 2.57)\\ 0.28\\ (2.33, 2.61)\\ \nu_3, \ \mathrm{cm}^{-1} \end{array}$	-171.9991	-172.0564	-172.1645	-171.9561	-172.1076
	-172.0033	-172.0585	-172.1638	-171.9559	-172.1068
	-172.0103	-172.0615	-172.1638	-171.9543	-172.1016
	-172.0155	-172.0633	-172.1641	-171.9511	-172.0934
	-172.0187	-172.0636	-172.1608	-171.9462	
			R's lar		
	A Casal	M. Land	C SIONS	arite at a	Basis II
$\begin{array}{c} 0.00\\ (2.38, 2.38)\\ 0.04\\ (2.36, 2.40)\\ 0.12\\ (2.32, 2.44)\\ 0.20\\ (2.28, 2.48)\end{array}$	-174.3817	-174.4196	-174.4676	-174.5137	
	-174.3854	-174.4206	-174.4686	-174.5134	-174.3866
	-174.3911	-174.4218	-174.4691	-174.5113	-174.3838
	-174.3943		eus:	-174.5068	
(2.20, 2.10) 0.28 (2.24, 2.52)	-174.3951	-174.4198	-174.4657	-174.5001	-174.3708
v <sub>3</sub> , cm <sup>-1</sup>		in Recovered	Chenner 45		
	BO <sub>2</sub> , SCF	BO2, R.S.*	BO <sub>2</sub> , CI	BO <sub>2</sub> , SCF	BO <sub>2</sub> , R.S.
	MO I	with MO I	with MO I	C∞v MO II	with MO II

R. S. - reference set.

 $BO_2^+$  ions. The  $v_3$  frequency was obtained by solving an ordinary vibration problem in a harmonic approximation.

As a starting point the symmetric linear configurations O—B—O at equilibrium geometries for the basis I ( $R_{B-O}=2.47$  a. u.) and the basis II ( $R_{B-O}=2.38$  a. u.) [<sup>3</sup>] were chosen. From these points the antisymmetric shifts of O atoms were made while  $\sum R_{B-O}$  was kept constant. The results are listed in the Table.

For asymmetric configurations the SCF solutions become unstable in a case of the  $X^2\Pi$  state of BO<sub>2</sub> as well as the  ${}^{4}\Delta$  state of BO<sub>2</sub><sup>+</sup>, and the total energy becomes lower than in appropriate symmetric points. Because of the multiconfigurational nature of this effect, the CI and CASSCF treatments will be considered.

When the basis I was used, SCF calculations of BO<sub>2</sub> in  $C_{\infty v}$  symmetry and without symmetry restrictions yielded the same values of total energies but different MOs. These solutions, as mentioned above, are unstable. The CI solution with excitations on  $\pi$ -orbitals only in the basis I yielded also an unstable one. Only including into the reference set excitations on  $\delta$ -orbitals enables to achieve stability. The CASSCF solution appears to be stable in the basis I, and yields a low

stretching frequency of BO<sub>2</sub>

a point and									
6	7	8	9	10	11				
and a second									
-172.1562	-171.5408	-172.0570	-172.1645	-172.1214	-172.1214				
-172.1567	-171.5443	-172.0288	-172.1624	—172.1213	-172.1213				
-172.1558	-171.5508	-172.0189	-172.1645	-172.1205	-172.1205				
-172.1545	are distant		AND	-172.1188	-172.1188				
	aT strig lou,	bits a self part		—172.1159	—172.1159				
A Paintafinies	nela colona color medita constanti		ind cipies for	1281.4	1281.4				
antin hard	a states			and an are	- Aller Mark				
-174.4408	-173.8254	-174.4212	-174.4529	-174.4726	-174.4726				
-174.4406	-173.8270	-174.4120	-174.4525	-174.4728	-174.4728				
-174.4392	-173.8318	-174.3932	-174.4484		A STATE AND A				
	-173.8360								
-174.4333	-173.8388								
1567.1					•				
BO <sub>2</sub> , CI	BO <sup>+</sup> <sub>2</sub> , SCF	BO <sub>2</sub> , R.S.	BO <sub>2</sub> , CI	BO <sub>2</sub> , CASSCF	BO <sub>2</sub> , CASSCF				
with MO II	MO III	with MO III	with MO III	with MO I	with MO II				

value of  $v_3 = 1281$  cm<sup>-1</sup> but it becomes unstable in the basis II. The CI solution with MOs from  $BO_2$  in the basis II yields also an unstable solution.

The SCF and CI treatments with MOs from  $X^{1}\Sigma$  state of BO<sub>2</sub> yield stable solutions in the basis I as well as in the basis II which confirm the low  $v_3$  value. The CASSCF treatment with these MOs in the basis I and II yields the same total energy values as with MOs from BO2, i. e. in the basis I it yields the stable solution, in the basis II - the unstable one. MOs from  $BO_{+}^{+}$  yield too large a  $v_{3}$  value.

#### REFERENCES

- Jug, K., Nanda, D. N. SINDO 1. III. Application to ground states of molecules containing fluorine, boron, beryllium and lithium atoms. Theor. Chim. Acta (Berlin), 1980, 57, N 2, 131—144.
   Saraswathy, V., Diamond, J. J., Segal, G. A. Theoretical calculation of the lowest electronic excited states of BO<sub>2</sub>. J. Phys. Chem., 1983, 87, N 5, 718—719.
   Lomp, P.-E., Simkin, V., Stepanov, N. Ab initio calculations of the ground and low-lying excited states of BO<sub>2</sub>. ENSV TA Toim. Keemia, 1983, 32, N 4, 286—288
- 286-288.

- 4. Johns, J. W. C. Absorption spectrum of BO<sub>2</sub>. Can. J. Phys., 1961, **39**, N 12, 1738-1768.
- 5. Серебренников Л. В. ИК-спектр поглощения молекулы ВО2, синтезированной в матрице из аргона. — Вестн. Моск. ун-та. Химия, 1975, 16, № 3, 363. 6. Коряжкин В. А., Саламова А. А. Частота  $v_3$  в основном электронном состоянии
- молекулы ВО<sub>2</sub>. Вестн. Моск. ун-та. Химия, 1975, 16, № 4, 487—489. 7. Коряжкин В. А., Матвеев В. К. Изотопные сдвиги в электронном спектре моле-
- кулы ВО2 и колебательные постоянные основного состояния. Вестн. Моск. ун-та. Химия, 1976, 17, № 4, 490—491. 8. Kawaguchi, K., Hirota, E., Yamada, C. Diode laser spectroscopy of the BO<sub>2</sub> radical.
- Vibronic interaction between the A<sup>2</sup>П<sub>u</sub> and X<sup>2</sup>П<sub>g</sub> states. Mol. Phys., 1981, 44, N 2, 509—528.
  9. Пупышев В. И., Симкин В. Я., Сафонов А. А., Хрустов В. Ф., Дементьев А. И.,
- Пункциев Б. н., Симкия Б. н., Сирков А. н., хрустов Б. ч., дежениев н. н., Степанов Н. Ф. Комплекс программ для неэмпирических расчетов молекул. Вестн. Моск. ун-та. Химия, 1983. 24, № 1, 38—39.
   Clementi, E., Roetti, E. Roothaan—Hartree—Fock atomic wave functions. Basis functions and their coefficients for ground and certain excited states of neutral and ionized atoms. Atom. Data and Nucl. Data Tables, 1974, 14, N. O. 107. 470.
- N 3. 177-478.
  11. Dunning, T. H. Gaussian basis functions for use in molecular calculations. I. Contraction of (9s5p) atomic basis sets for the first row atoms. J. Chem. Phys., 1970, 53, N 7, 2823-2833.
- 12. Пупышев В. И., Симкин В. Я., Сафонов А. А., Дементьев А. И. Отбор конфигураций в методе конфигурационного взаимодействия. — Вестн. Моск. ун-та. Химия, 1984, 25, № 2, 161—164.

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Received Jan. 21, 1985